Heat Transfer, Fluid Flow, and Metallurgical Transformations in Arc Welding: Application to 16MND5 Steel

F. Roger, A. Traidia, B. Reynier

Abstract — Arc welding creates a weld pool to realize continuity between pieces of assembly. The thermal history of the weld is dependent on heat transfer and fluid flow in the weld pool. The metallurgical transformation during welding and cooling are modeled in the literature only at solid state neglecting the fluid flow. In the present paper we associate a heat transfer – fluid flow and metallurgical model for the 16MND5 steel. The metallurgical transformation model is based on Leblond model for the diffusion kinetics and on the Koistinen-Marburger equation for Martenise transformation. The predicted thermal history and metallurgical transformations are compared to a simulation without fluid phase. This comparison shows the great importance of the fluid flow modeling.

Keywords — Arc welding, Weld pool, Fluid flow, Metallurgical transformations.

I. INTRODUCTION

Gas Tungsten arc welding process (GTAW) realizes metal continuity by fusion of assembly. The resulting weld area is characterized by a fusion zone and a heat affected zone where solid-solid metallurgical transformations take place during welding and cooling steps.

Considering metallurgical transformations in welding simulation is necessary because they influence residual stresses and distortions. Indeed, thermal and mechanical material properties are fraction of phase dependent: for example thermal expansion coefficient of Aустenite is equal to $23.5\text{e-6}^\circ\text{C}^{-1}$ while its value is $15\text{e-6}^\circ\text{C}^{-1}$ for Ferrite [1]. Moreover, for steel like 16MND5 (A508 CI3), inelastic strains like transformation Induced Plasticity result of phases transformation under mechanical stresses.

Modeling of metallurgical transformation during welding or heat treatment has been extensively studied [2]-[5], [8]. In these previous works, phase change takes place at solid state between room temperature and austenization temperature (1200$^\circ$K). The cooling conditions are described using CCT diagram. In welding, fusion and fluid flow in the weld pool lead to quite different thermal histories because of the convective heat transfer in the weld pool. The heating and cooling speed are very high, several hundreds of degree per second. Therefore to predict the metallurgical transformations during welding, we consider the fluid flow in the weld pool to have realistic heating and cooling conditions.

A heat transfer, fluid flow and metallurgical transformation model have been developed. The weld pool is considered as an incompressible and viscous fluid governed by Navier-Stokes equation. Solid metallurgical transformations are based on a 16MND5 steel CCT diagram and metallurgical kinetics are based on Leblond model [2] for diffusion equations. Martensite evolution is described by Koistinen Marburger equation [6] with a minimum cooling speed condition. Koistinen-Marburger equation coefficients have been adjusted to CCT diagram with a bainite phase fraction dependence. Metallurgical predictions will be compared to a thermo-metallurgical simulation without fluid phase.

II. MATHEMATICAL FORMULATION OF THE PROBLEM

The model is divided into two parts. The first one is a heat transfer and fluid flow simulation. The second part evaluate fraction of each metallurgical phase from the thermal history given by the previous model.

A. Heat transfer and fluid flow in the weld pool

In arc welding, arc heats the top surface of the assembly during a few seconds to create a weld pool. This surface heat source can be approximate by a Gaussian heat distribution according the following equation:

$$ q_{weld} = \frac{3\pi U I}{\alpha_a} \exp\left(-\frac{r^2}{\alpha_a^2}\right) $$  \hspace{1cm} (1)
Where $q_{\text{welding}}$ is the surface heat flux (W.m$^{-2}$), $U$ is the welding voltage (V), $I$ is the current (A), $\eta$ is the arc efficiency (0.68 for GTAW), $r_i$ is the arc radius (m).

The other sides of the assembly are submitted to air convection. This heat distribution creates thermal gradient at the top surface. The surface tension of stainless steel is very sensitive to temperature. As a consequence, surface tension gradients occur. In the present study, we consider that the temperature coefficient of surface tension is temperature dependent with a constant sulfur content of 20 ppm. Generally, this coefficient is negative for pure metals but the presence of sulfur in stainless steel can make its value positive.

The following expression is used for the temperature coefficient of surface tension [7]:

$$\frac{\partial \gamma}{\partial T} = \left( A_y - R_g \Gamma_s \ln \left( 1 + K \alpha \right) \right) - \frac{K(T) \alpha}{1 + K(T) \alpha} \frac{\Delta H_0}{T} \quad \left( \frac{\partial \gamma}{\partial T} \right)$$  \hspace{1cm} (2)

$$K(T) = k_1 \exp \left( \frac{-\Delta H_0}{R_g T} \right) \quad \left( K(T) \right)$$  \hspace{1cm} (3)

Where $\gamma$ is the surface tension, $A_y$ is a constant, $R_g$ is gas constant, $\Gamma_s$ is surface excess at saturation, $\alpha$ is activity of sulfur, $\Delta H_0$ is standard heat of adsorption, $k_1$ is the entropy factor. The values of these parameters are presented in table 1 in appendix.

Gradients of surface tension at the top surface of the weld pool lead to shear stresses according to the Marangoni effect. This shear stress produces fluid flow in the weld pool as the liquid metal is viscous.

The Marangoni shear stress boundary condition, for an axial symmetric heating, is described by the following equation:

$$\frac{\partial u}{\partial z} = f_l \frac{\partial \gamma}{\partial T} \frac{\partial T}{\partial r} \quad \left( \frac{\partial u}{\partial z} \right)$$  \hspace{1cm} (4)

Where $r$ is the radius from the arc center, $z$ is the height, $u$ is the radial velocity, $\mu$ is the dynamic viscosity, $\gamma$ is the surface tension, $f_l$ is the liquid fraction (equal 1 at the fluid free surface and decreasing from 1 to 0 for solid state in the mushy zone of the weld). The normal fluid velocity is set to zero.

The governing equation for the heat transfer and fluid flow are expressed as follows:

Conservation of mass:

$$\nabla \cdot \vec{v} = 0 \quad \left( \nabla \cdot \vec{v} \right)$$  \hspace{1cm} (5)

Conservation of momentum:

$$\rho \left( \frac{\partial \vec{v}}{\partial t} + \vec{v} \cdot \nabla \vec{v} \right) = -\nabla p + \mu \nabla \left( \nabla \vec{v} + \nabla \vec{v}^T \right) + \vec{F}_b$$  \hspace{1cm} (6)

Where $\vec{v}$ is the fluid velocity, $\vec{F}_b$ is the buoyancy force (Boussinesq approximation) and is expressed by:

$$\vec{F}_b = \rho_0 (1 - \beta (T - T_{ref}) \vec{g}) \quad \left( \vec{F}_b \right)$$  \hspace{1cm} (7)

Where $\rho_0$ is the density at $T = T_{ref}$, $\beta$ is the thermal expansion coefficient.

Energy conservation:

$$\rho C_p \frac{\partial T}{\partial t} + \rho C_v \vec{v} \cdot \nabla T = \nabla \left( \lambda \nabla T \right) + \rho L_f \frac{df_l}{dt} \quad \left( \rho C_p \right)$$  \hspace{1cm} (8)

Where $f_l$ has been defined in the Marangoni equation (2), $L_f$ is the latent heat of fusion. The main parameters for the heat and fluid flow simulation are presented in table 1.

**B. Metallurgical transformations**

Welding is associated to anisothermal metallurgical transformation with high heating and cooling speeds. In the case of 16MND5 steel, kinetics corresponds to diffusion and martensitic transformations. J.B. Leblond [2] uses fraction of each phase $z_i$ as variable and proposes to describe all this kinetics using Ordinary Differential Equations (ODE). The Martensitic transformation is usually described by the Köstinen Marburger equation [6].

**B.1 Leblond-Devaux’s model**

The evolution of fraction of phase is governed by temperature for Austenite, Ferrite-Perlite and by temperature and cooling rate for bainite. The Austenite grain size has also a great influence as it modify the Continuous Cooling Diagram of the steel. This parameter can be add as a variable in the coefficient of the transformation equations.

The general form of the evolution equation is the following ODE:

$$\dot{z} = \frac{z_{eq}(T) - z}{\tau(T)} \quad \left( \dot{z} \right)$$  \hspace{1cm} (9)

Where $z_{eq}$ is the fraction of phase at the thermodynamics equilibrium for a constant temperature $T$, so that when $z = z_{eq}(T)$, the equation (9) doesn’t evolve, $\tau$ is a characteristic time to reach $z = z_{eq}$ from $z = 0$. The temperature dependence of $z_{eq}$ is obtained at very low heating rates. For Martensite transformation, J.B. Leblond use the same ODE to keep the same framework but with a shorter characteristic time.
The figure 1 shows the qualitative predictions of equation 9.

![Figure 1: Temperature dependence of z according to the heating rate. For very low heating rate, z(t) follows \( z_{eq}(T(t)) \) [Leblond 1984].](image)

At each time, the sum of all the fractions of phase is equal to 1. Then, \( z_1 + z_2 + z_3 + z_4 = 1 \).

Where \( z_1 \) is fraction of ferrite-perlite, \( z_2 \) is fraction of bainite, \( z_3 \) is fraction martensite and \( z_4 \) is fraction of austenite. This condition can be written in time derivative again:

\[
z_1 + z_2 + z_3 + z_4 = 0 \tag{10}
\]

Leblond writes equation (9) again in a more suitable form for generalization, for a unique \( 1 \to 2 \) transformation, equation (9) becomes:

\[
z_2 = \frac{z_{eq}(T)}{\tau(T)} - z_2 + \frac{z_{eq}(T)}{\tau(T)} - \frac{z_{eq}(T)}{\tau(T)} z_1
\]

\[
z_2 = \frac{z_{eq}(T)}{\tau(T)} z_2 - \frac{z_2}{\tau(T)} + \frac{z_{eq}(T)}{\tau(T)} z_1
\]

\[
z_2 = \frac{z_{eq}(T)}{\tau(T)} z_2 - \frac{z_2}{\tau(T)} + \frac{z_{eq}(T)}{\tau(T)} z_1
\]

\[
z_2 = k_{12}(T) z_1 - l_{12}(T) z_2 \tag{11}
\]

With \( k_{12}(T) = \frac{z_{eq}(T)}{\tau(T)} \) and \( l_{12}(T) = 1 - \frac{z_{eq}(T)}{\tau(T)} \)

For a unique \( 1 \to 2 \) transformation, the equation (10) gives:

\[
z_1 = -z_4 = -k_{12}(T) z_1 + l_{12}(T) z_2 \tag{12}
\]

If \( k_{12}(T) z_1 - l_{12}(T) z_2 > 0 \)

For the \( 2 \to 1 \) transformation, we have:

\[
z_1 = k_{21}(T) z_2 - l_{21}(T) z_1 \tag{13}
\]

\[
z_2 = -k_{21}(T) z_2 + l_{21}(T) z_1 \tag{14}
\]

If \( k_{21}(T) z_1 - l_{21}(T) z_2 > 0 \)

There is no transformation \( z_1 = z_2 = 0 \) if

\[
k_{12}(T) z_1 - l_{12}(T) z_2 \leq 0 \text{ and } k_{12}(T) z_1 - l_{12}(T) z_2 \leq 0.
\]

For a good description of the Austenite/Bainite transformation which takes place in a wide range of cooling conditions, Leblond add a cooling rate dependence for the \( k_{12} \) and \( l_{12} \) functions and proposed the following factorized forms:

\[
k_{12}(T, \dot{T}) = f_{12}(T) * h_{12}(\dot{T}) \tag{15}
\]

\[
l_{12}(T, \dot{T}) = f_{12}(T) * h_{12}(\dot{T}) \tag{16}
\]

### B.2 Martensite transformation

Martensite phase appears at very high cooling rates far from thermodynamic equilibrium. The phase change transformation is so quick (about the speed of sound in a metal) that we can consider it as instantaneous. That’s why Koistinen and Marburger have developed a transformation equation for Martensite without time dependence. In the high cooling rate conditions, the Martensite fraction is dependent on the temperature shift related to the Martensite start temperature (Ms). The Koistinen-Marburger equation is as follow:

\[
z_3 = z_4 * \left( 1 - \exp \left( \beta * [M_s - T] \right) \right) \tag{17}
\]

Where \([.]^+\) is the positive part operator, Ms is the Martensite start temperature, \( \beta \) is a coefficient adjusted such that when \( T = M_f \), \( z_3 = 0.99 * z_4 \).
B.3. Metallurgical transformation in 16MnD5 steel

Leblond identify the functions k(T) and l(T) of the transformation equations using a CCT diagram for the 16MnD5 steel (see Fig. 2). All the identified coefficients for the transformations are presented in table II.

Where \([\text{cond.}] = 1\) if the condition is realized and zero otherwise. \(T_m^\text{p}\) is the minimum cooling speed to form Martensite. In the 16MnD5 CCT diagram it corresponds to 1°C/s.

The resulting Martensite transformation equation is:

\[
z_3 = z_4 \left(1 - \exp \left(\beta(z_2) \left[Ms(z_2) - T\right]^{-1}\right)\right) [\gamma > 0]
\]

(20)

The coefficient \(\beta\) is chosen dependent of the bainite fraction because the shift between Martensite start and Martensite finish temperature is not constant in the CCT diagram. We use the following formula:

\[
\beta(z_2) = \frac{\ln 0.01}{Ms(z_2) - Mf}
\]

(21)

Where Martensite finish temperature is fixed to 473°C. The value 0.01 is fixed to have \(1 - \exp \left(\beta(z_2) \left[Ms(z_2) - T\right]^{-1}\right)\) = 0.99 i.e. when \(T\) is equal to \(Mf\), 99% of Austenite is transformed into Martensite.

III. APPLICATION: MODELING OF SPOT ARC WELDING

Previous Heat transfer, fluid flow and metallurgical transformations models have been applied to simulate the arc spot welding of a 16MnD5 steel disk (thickness 6 mm and radius 20mm). The arc heating is maintained constant during 5 seconds at the top center of the disk followed by air cooling during 15s. As heating and clamping conditions are axial symmetric, the simulation is done in 2D axial symmetric conditions. Simulations are realized using the finite element code Comsol Multiphysics®.

As metallurgical transformation kinetics equations are not present in the Comsol modules, we have developed our own toolbox using the Partial Differential Equation module. The simulations of weld pool evolution and the metallurgical transformations are described in the next sections.

A Heat transfer and fluid flow in the weld pool

The following figure shows fluid flow and temperature field at the end of the heating time.
The Marangoni effect induced a clockwise vortex and then arc heat is convected from the center of the top surface to the radial direction. The resulting weld pool is wide (r=3.4mm) and weakly penetrated (0.8mm). The maximum radial velocity at the top surface is 0.18m/s near the outer edge of the weld pool.

The thermal history of vertical points at different height (0 to 6mm from the bottom of the disk) at r=0 is presented in figure 4.

The maximum cooling rate for these different height (0-1, 1-2, 2-3, 3-4, 4-5mm) is respectively from the bottom to the top (-110, -120, -145, -210, -340, -540, -1300°C/s). Please note that the 6mm height point presents a singularity and is not considered.

**B Metallurgical transformations in the welded zone**

The simulation start with a cold disk having a fraction of ferrite-perlitic $z_1=0.4$ and bainite $z_2=0.6$.

At the end of arc heating (t=5s) the austenitic phase distribution is presented in figure 5 (fraction of $z_2$). From this austenite phase, martensite and bainite will be formed during cooling. The Figure 6 shows the final Martensite phase distribution after cooling (t=20s).
Martensite is formed after cooling in the fusion zone with a little bit of Bainite (fig 7a). In the heat affected zone, (fig 7b) we have a mixture of Bainite, Ferrite and Martensite. The sum of phase fractions is closed to 1 (equation 10), so this constraint is well respected in our model.

C Comparison with a thermo-metallurgical simulation

To study the importance of considering the weld pool in the modeling, we compare prediction with and without liquid phase. The weld pool, with convective heat transfer, increases the heat flow rate inside the workpiece. The thermal history of vertical points on the axis at different heights (0 to 6mm from the bottom of the disk) is shown in figure 8. It permits a comparison with the fully coupled model results (fig 4).

The final phase is made of quite Martensite while it is a mixture of Bainite-Martensite and Ferrite for the fluid flow model.

IV. CONCLUSION

The prediction of metallurgical transformations in welding, which is controlled by the thermal history, necessitates considering the fluid flow modeling in the weld pool. Usually, the metallurgical consequences of welding are modeled using heat conduction model like in the commercial software Sysweld®.

APPENDIX

TABLE I

<table>
<thead>
<tr>
<th>Symbol</th>
<th>Quantity</th>
<th>Conversion from Gaussian and CGS EMU to SI*</th>
</tr>
</thead>
<tbody>
<tr>
<td>( \rho )</td>
<td>Density</td>
<td>6080 to 7272 kg.m(^{-3})</td>
</tr>
<tr>
<td>( c_p )</td>
<td>Specific heat</td>
<td>510 to 796 J.kg(^{-1}).K(^{-1})</td>
</tr>
<tr>
<td>( k )</td>
<td>Thermal conductivity</td>
<td>37.7 to 26.9 W.m(^{-1}).K(^{-1})</td>
</tr>
<tr>
<td>( \mu )</td>
<td>Dynamic viscosity</td>
<td>0.05 kg.m(^{-1}).s(^{-1})</td>
</tr>
<tr>
<td>( \sigma )</td>
<td>Electrical conductivity</td>
<td>7.7 x 10(^{-3}) (\Omega) m(^{-1})</td>
</tr>
<tr>
<td>( T_l )</td>
<td>Liquidus temperature</td>
<td>1723 K</td>
</tr>
<tr>
<td>( T_s )</td>
<td>Solidus temperature</td>
<td>1673 K</td>
</tr>
<tr>
<td>( \alpha_s )</td>
<td>Sulfur activity</td>
<td>0.001 wt%</td>
</tr>
<tr>
<td>( R_g )</td>
<td>Gas constant</td>
<td>8314 J.kg(^{-1}).mole(^{-1}).K(^{-1})</td>
</tr>
<tr>
<td>( \Delta H_f )</td>
<td>Latent heat of fusion</td>
<td>-1.66 x 10(^{8}) J.kg(^{-1}).mole(^{-1})</td>
</tr>
<tr>
<td>( \Gamma_s )</td>
<td>Surface excess of sulfur at saturation</td>
<td>1.3 x 10(^{-4}) J.kg(^{-1}).mole(^{-1}).m(^{-2})</td>
</tr>
<tr>
<td>( \gamma_n )</td>
<td>Surface tension at pure metal</td>
<td>1.943 N.m(^{-1})</td>
</tr>
<tr>
<td>( k_c )</td>
<td>Entropy factor</td>
<td>3.18 x 10(^{-3}) N.m(^{-1}).K(^{-1})</td>
</tr>
<tr>
<td>( A_s )</td>
<td>Constant in surface tension gradient</td>
<td>4.3 x 10(^{-6}) N.m(^{-1}).K(^{-1})</td>
</tr>
</tbody>
</table>
TABLE II
PARAMETERS USED FOR THE COEFFICIENTS
OF FRACTION PHASE IN THE LEBLOND MODEL

<table>
<thead>
<tr>
<th>Temp. °C</th>
<th>0</th>
<th>730</th>
<th>750</th>
<th>770</th>
<th>790</th>
<th>810</th>
</tr>
</thead>
<tbody>
<tr>
<td>k_{14}</td>
<td>0</td>
<td>0</td>
<td>0.22</td>
<td>0.53</td>
<td>1.05</td>
<td>2.02</td>
</tr>
<tr>
<td>k_{34}</td>
<td>0</td>
<td>0</td>
<td>0.97</td>
<td>0.94</td>
<td>0.87</td>
<td>0.76</td>
</tr>
<tr>
<td>l_{14}</td>
<td>1</td>
<td>1</td>
<td>0.97</td>
<td>0.94</td>
<td>0.87</td>
<td>0.76</td>
</tr>
<tr>
<td>l_{34}</td>
<td>0.45</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
</tr>
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</table>

Ferrite-pearlite transformation

<table>
<thead>
<tr>
<th>Temp. °C</th>
<th>0</th>
<th>600</th>
<th>800</th>
<th>1000</th>
</tr>
</thead>
<tbody>
<tr>
<td>k_{42} (s⁻¹)</td>
<td>1e-4</td>
<td>1e-4</td>
<td>1.8e-3</td>
<td>0</td>
</tr>
<tr>
<td>l_{42} (s⁻¹)</td>
<td>0</td>
<td>2e-4</td>
<td>2e-3</td>
<td>2e-3</td>
</tr>
</tbody>
</table>

 Bainite transformation

<table>
<thead>
<tr>
<th>Temp. °C</th>
<th>340</th>
<th>350</th>
<th>450</th>
<th>550</th>
<th>1000</th>
</tr>
</thead>
<tbody>
<tr>
<td>h_{12} (s²)</td>
<td>0</td>
<td>0.014</td>
<td>0.067</td>
<td>0</td>
<td>0</td>
</tr>
<tr>
<td>h_{32} (s²)</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0.067</td>
<td>0.067</td>
</tr>
</tbody>
</table>

Cooling rate (°C s⁻¹)

| h_{42} | 0.2 | 1 | 1.3 | 0.22 | 0.1 | 0.0044 |

REFERENCES


